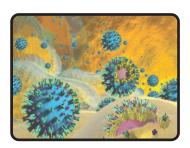


EduHeal Foundation Nationwide Biotechnology Olympiad

and other

National/International Biotechnology Olympiads/Talent Search Exams.



Class-11

EtG-BOOKS

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SYLLABUS GUIDELINES*

Based on CBSE, ICSE & GCSE Syllabus & NCF guidelines devised by NCERT

Key Topics in Mathematics for Class XI and XII

I. ALGEBRA

Sets, Relations and Functions, Complex Numbers, Matrices and Determinants, Quadratic Equations, Permutation and Combination, Binomial Theorem and its Applications, Sequences and Series

II. CALCULUS

Differential Calculus, Integral Calculus, Differential Equations

III. TWO AND THREE DIMENSIONAL GEOMETRY

Two dimensional Geometry, The straight line and pair of straight lines, Circles and system of Circles, Conic Section, Three dimensional Geometry

IV. VECTORS

Vector Algebra

V. STATISTICS

Measures of Central Tendency and Dispersion, Probability

VI. TRIGONOMETRY

VII. STATICS AND DYNAMICS

Statics, Dynamics

Key Topics in Physics for Class XI and XII

- 1. Units and Measurement
- 2. Description of Motion in one dimension
- 3. Description of Motion in Two and Three dimensions
- 4. Laws of Motion
- 5. Work, Energy and Power
- 6. Rotational Motion and Moment of Inertia
- 7. Gravitation
- 8. Properties of Matter
- 9. Oscillations
- 10. Waves
- 11. Heat and Thermodynamics
- 12. Transference of heat
- 13. Electrostatics
- 14. Current Electricity
- 15. Thermal and Chemical Effects of currents
- 16. Magnetic Effects of Currents
- 17. Magnetostatics
- 18. Electromagnetic Induction and Alternating Currents
- 19. Ray Optics
- 20. Wave Optics
- 21. Electromagnetic waves
- 22. Electrons and Photons
- 23. Atoms, Molecules and Nuclei
- 24. Solids and Semi-Conductor Devices

Class - 11 v

Key Topics in Chemistry for Class XI and XII

- 1. Atoms, Molecules and Chemical Arithmetic
- 2. Elements, their Occurrence and extraction
- 3. States of Matter Gaseous state
- 4. Atomic Structures Constituents of the atom
- Chemical Families Periodic Properties
- 6. Chemical Bonding and Molecular structure
- 7. The Solid State
 The Gaseous state
- 8. Solutions
- 9. Chemical Energetics and Thermodynamics
- 10. Chemical Equilibrium
- 11. Redox Reactions and Electrochemistry
- 12. Rates of Chemical Reactions and Chemical Kinetics
- 13. Chemistry of Hydrocarbons
- 14. Purification and Characterisation of Organic Compounds
- 15. Organic Chemistry Based on Functional Group
- 16. Organic Chemistry Based on Functional Group II
- 17. Organic Chemistry Based on Functional Group-II
- 18. Chemistry of Non-metals
- 19. Chemistry of Non-metals II
- 20. Chemistry of lighter Metals
- 21. Heavy Metals
- 22. Chemistry of Representative Elements
- 23. Transition Metals including Lanthanides
- 24. Coordination Chemistry and Organo Metallics
- 25. Nuclear Chemistry
- 26. Synthetic and Natural Polymers
- 27. Surface Chemistry
- 28. Bio Molecules and Biological Processes
- 29. Chemistry in Action

Key Topics in Biology for Class XI and XII

- 1. General Biology
- 2. Systematics and Classification
- 3. Animal Kingdom
- 4. Plant Kingdom
- Cell Biology
- 6. Genetics
- 7. Human Biology
- Angiosperm Botany
- 9. Ecology and Environment
- 10. Application of Biology
- 11. Evolution



For Maths, Physics & Chemistry please refer to Maths and Science Workbooks.

INTERNATIONAL BIOLOGY
OLYMPIAD MEDALISTS

= MAKING INDIA PROUD=

Edusys: How did you prepare for the International Biology Olympiad? Is any special preparation required for it?

Anurag Shahi: For the International round of the exam, we were prepared by the teachers at HBCSE. They guided us and provided us books to study. The training camp was excellent and exhaustive and one can really learn a lot of things which help him even in further studies. As the questions are quite different from the one usually asked in other exams such as PMT, some special preparation is required. One should solve previous years



(L - R) Devanshu Bansal (Silver medalist), Anurag Chahal (Silver medalist), Kalyani Kansal (Gold medalist) and Anurag Shahi (Silver medalist).

papers to know the pattern of questioning and study accordingly. Practicals are very-very important part and hence it is an area to concentrate.

Devanshu Bansal : Concept clearing and application of the basic knowledge and logic is important. Practical component has quite a large weightage, hence equal stress on practicals and theory is required. Training camp gives good practice.

Anurag Chahal: There is no special preparation required for the International Biology Olympiad. While studying, you should keep in mind that you have to clear your concepts. Whatever you read, you should be 100% clear about the concepts.

Kalyani Kansal: Initially we had an Orientation-cum-selection camp at Mumbai from 3 June to 11 June. After that we had a Pre-Departure Camp in July, which was for 14 days. During these camps, we were oriented with the lab practices, we did dissections, general slide preparations, biochemical and genetics experiments etc. And besides that we did old theory papers of Indian National Biology Olympiad (INBO) and IBO. So that was the preparation we did. As such, Olympiad is not something for which you can prepare really. Its what all

you have gathered during the past 2 years. There is no specific "pattern" for which one can prepare. One has just to be clear about the different ideas, the very basic facts (Eg. the direction of DNA replication etc.) and keep one's mind open as he/she goes to take the exam.

Edusys: What books you referred to?

A.S.: 'Campbell' is a must for the International part. It is the *bible* for IBO. Other useful books are Guyton, Biological Sciences (Cambridge University Press), and Lehninger. (These books are used as physiology & biochemistry text books in 1st year MBBS)

D.B: Among books, Campbell is the most recommended. It gives good practise.

A.C.: You should go through Biology by Campbell & Reece if you really want to score good in IBO. Probably I did not get a gold medal because I did not go through this book. NCERT books are required only to clear the first round but 2nd round onwards only Campbell helps and so does Biological Sciences - Cambridge University Press by Soper *et al.* Lehninger is also good book for biochemistry.

K.K.: Neil A. Campbell's Biology.

Edusys: What was your experience when you wrote the International Biology Olympiad?

A.S.: While writing any exam my mind just goes blank, I mean blank to everything except the questions and options. I felt nervous & tensed just before the exam started. Once the exam started, I just felt like answering the questions. At that time their is no room for nervousness, any sort of doubt or anything else. You just have to concentrate hard on the questions, shut your mind to all other thoughts and give your best shot.

D.B.: The IBO exam was very good. The practicals were skilfully designed and it required practice to do them properly. But after the training camp, they did not seem that difficult. The theory paper was quite easy.

A.C.: I was quite excited & a bit nervous as I was representing India at the International level and all the eyes were set on me and the medal and glory I was going to get for my country depended on my performance in those few hours.

K.K.: Obviously it was the experience of a lifetime. I mean, think about it. With a mere Rs 50 you are trained with the best teachers of the country, meet new people, live in luxury, go abroad etc. As for the exam, it didn't throw any surprises as such, except for the fact that it was a rather simple one as compared to INBOs and some other IBOs.

Edusys: What was the pattern of exam and what was the question paper like? Can you recall any question which appeared in the exam?

A.S.: The pattern was different from the usual exams. It was mostly applied but stress was laid on little details which sometimes skip notice. If you have gone through the text thoroughly with understanding, it is easy. The were questions like filling tables, matching columns etc. These look easy but you need good solid basics and thinking ability to get through.

D.B: The exam had 2 parts - theory and practical. The practical part had 4 topics - Genetics, Biochemistry, Plant Sciences and Animal Sciences with 1½ hrs. for each topic. The theory included all the different aspects of biology (like Cytology, Ethology, Ecology, Genetics and Evolution, Botany, Zoology, Systematics). The theory exam had 2 parts, one with MCQs and the other with match the following, fill in the blank type.

For example, in Genetics practicals, there was DNA extraction, in Biochemistry, there was colorimetry & sectioning in Botany.

A.C. The pattern of the exam was as if they wanted to test your basic knowledge. They did not expect you to know details of each organism or each unique phenomenon, but they wanted that you to know the basic concepts regarding very elementary processes like glycolysis. But you should be clear enough about what you know. Question paper seemed quite easy but was tricky which would include some hidden hints which had to be seen before final attempt.

K.K.: As for the theory exam, there are objective questions mainly (except a few related to cladograms), most of them are multiple choice, others are matches (hardly any factual matches) etc. As per the IBO, the exam shouldn't have more than 25% factual questions. The practicals are like, we are given a detailed set of instructions, and we have to follow these and be patient and accurate. One has to have plenty of presence of mind for the practicals. Then in the follow-up questions we are asked to interpret the results and predict results in case of any change in the experimental setup.

Edusys: Students who appeared in the previous International Biology Olympiad said that ethology is an important chapter for International Biology Olympiad. What do you say?

A.S.: Yes, Ethology is very very important.

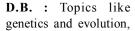
D.B.: Yes, Ethology can have questions which require logic and application of the basic knowledge about the patterns of behaviour of the various animals.

A.C.: Yes, ethology is indeed an important part of Ecology as per IBO syllabus. In Biology by camphell, the details given are enough and this book demartates the syllabus for International Biology Olympiad.

K.K.: Yes it is. But its not a big deal. Once you get acquainted with some basic terms like habituation, altruism etc., its all about having some presence of mind. But sometimes there are certain important examples which might be asked about, so in that case, one ought to refer to Campbell.

Edusys: Which topics are important for International Biology Olympiad?

A.S.: The topics which are generally neglected for PMT's and some entirely new topics like Ethology, Ecology, Biostatistics, Biosystematics, Probability analysis, Genetics are important.





(L - R) Anurag Shahi, Anurag Chahal, Devanshu Bansal and Venkatraman (Ex-IBO gold medalist) with Dr. Sandeep Ahlawat (second from right)

Biostatistics are quite important for International Biology Olympiad.

A.C.: Biostatistics, Genetics- in a little detail and evolution, Biosystematics, Behavioural Ecology and Ethology are important topics for International Biology Olympiad.

K.K.: Genetics, Plant (especially growth of plants: effects of light etc.) and Animal Physiology, Systematics, Ecology and Ethology are important. For the botany practicals, plant anatomy and systematics are very important.

Edusys: What would you suggest to those aspiring to write Olympiad exams at both National and International levels?

A.S.: I would adirce them to have faith in themselves and their teachers. They shouldn't follow the crowd but seek their own path. Everyfore is different and what suiter one isn't necessarily suitable for everyone else. So just work hard and don't lose heart.

D.B.: The Olympiad is an excellent opportunity for getting to improve your concepts about the various aspects of biology as well as meeting with the students from other countries. Those who plan to presume a research career get an excellent opportunity by participating in International Biology Olympiad.

A.C.: I would suggest that students should have faith in their ability and should work hard and should be determined enough together through PMTS as well as Olympiads. Clear your concepts will and don't cram. They should not indulge many activities which is oriented towards their ultimate goal atleast for these two crucial carrier-making years.

K.K.: Read from good books. And read not because you want to crack a certain exam, but because you really care about the subject.



EXERCISE-1 1. Mesosomes - a cell wall associated structures in bacteria are concerned with the process of (a) Septum formation (b) Respiration (c) Digestion (d) Both (b) and (c) 2. In bacteria, besides a cell wall there is a capsule surrounding cell wall commonly made up of (a) Polypeptides (b) Polysaccharides (c) Monosaccharides (d) Polyglycogens 3. Gram positive bacteria retain Gram's stain owing to the high percentage of than Gram negative bacteria. (a) Peptidoglycan (b) Lipopolysaccharides (c) Lipoproteins (d) Proteins 4. Species can be defined as a (a) Group of bacteria similar in characters and essentially fertile selfbreeders. (b) Group of bacteria forming similar types of colonies. (c) Group of similar bacteria (d) Group of bacteria that coexists together and form colony. 5. One of the following is a Gram negative bacteria, it is (a) Spirilla (b) Ervsiphilothrix (c) Corynebacteria (d) Lactobacilli Which of the following is not cocci bacteria? (b) Streptococcus (a) Streptomyces (c) Strephylococcus (d) Sarcina 7. In which of the following group chain of conidia grows on aerial hyphae (a) Streptomyces (b) Nocardia (d) All of the above (c) Actinomyces The basic difference between Actinomyces and Nicardia group of bacteria is that (a) Former is filamentous bacteria while latter is spiral. (b) Former is a anaerobic while latter is aerobic. (c) Former is Gram negative while latter is Gram positive. (d) Former is aerobic while latter is anaerobic 9. Actinomyces, a subgroup of filamentous bacteria is (a) Gram positive (b) Gram negative

(a) Rhizobium

(c) Leptothrix ochracae

19. Growth may be defined in simple words as a/an

€tG Olympiad Explorer (c) Fuelgen positive (d) Fuelgen negative 10. The cells of gram positive bacteria have (a) Multiple layers containing high amount of peptidoglycan (b) Periplasmic space (c) Large amounts of teichoic acids (d) Layers having less amount of peptidoglycan 11. Which of the following is true for photoorganotrophs (a) $NH_2 + 3/2O_2 \xrightarrow{Nitrosomonas} NO_2 + H^+ + H_2O$ (b) $2H_2S + CO_2 \xrightarrow{Light} CH_2O + H_2O + 2S$ (c) $2CH_3CHOHCH_3 + CO_2 \xrightarrow{Light} CH_2O + 2CH_3CCH_3 + H_2O$ (d) $2HNO_2 + O_2 \xrightarrow{Nitrosomonas} 2HNO_3 + Energy$ 12. The amount of oxygen required by organic matter in a sample of water for its oxidation by a strong chemical oxidant is called as (a) BMR (b) CD (c) BOD (d) COD 13. Enterobacter helps in production of from pyruvic acid. (a) Acetone (b) Butanediol (c) Ethanol (d) Propanol 14. Lactic acid is produced from pyruvic acid at industrial level by the help of (b) Lactococcus (a) Staphylococcus (c) Streptococcus (d) All of the above. 15. Which of following bacteria are responsible for conversion of pyruvic acid to propionic acid through oxaloacetic acid formation (a) Enterobacter (b) Propionobacterium (c) Lactobacillus (d) Clostridium **16.** Thermophiles are bacteria which can grow in places of high temperature. Which one is a thermophile? (a) Pseudomonas spp. (b) Rhizobium spp. (c) Chlostridium spp. (d) Thermophilus aquaticus 17. Which one of the following bacteria oxidizes molecular hydrogen in order to form water in soil? (a) Bacillus pentotrophus (b) Rhizobium noduliformis (c) Albugo candida (d) All of the above 18. One of the following bacteria oxidizes ferrous compounds into ferric compounds find it out

(b) Bacillus pentotrophus

(d) Bacillus thermophilus

(a) Reversible change in shape and size (b) Irreversible change in mass and volume (c) Reversible change in mass and volume (d) Irreversible change in few systems only. 20. Polymetaphosphate, a storage granule acts as a source of (a) Phosphate (b) Sulphate (c) Sulphur (d) Phosphoric acid 21. The bacteria, Bacillus mesenterius is used for the production of & enzymes. (a) Pectinase, cellulase (b) Amylase, protease (c) Polymerase, invertase. (d) Cellulase oxide, carboxylase 22. In alcoholic fermentation, there are two species responsible for top and bottom fermentation. They are (a) S. cerevisal and S. saccharum (b) S. carlsbergensis and S. cerevisal (c) Saccharomyces cerevisae and S. carlsbergensis (d) S. carlsbergensis and S. saccharum 23. Fuel oil, a fermentation product is the mixture of (a) Isoamyl + Methanol (b) Methanol + Butanol (c) Ethanol + Methanol (d) Isoamyl Alcohol + Butyl Alcohol 24. Bacterial conjugation a mean of transferring genetic material from one cell to other was first described by (a) Watson and Crick (b) Leaderberg and Tautum (c) Leaderberg (d) Tautum 25. The most common type of reproduction in bacteria is (a) Budding (b) Binary fusion (c) Binary fission (d) Sexual reproduction 26. The microorganism responsible for major alcoholic fermentation are (a) Pseudomonas spp. (b) Lactobacillus spp. (c) Saccharomyces spp. (d) All of the above 27. Clostridium spp. are good fermenter responsible for the formation of (a) n-butanol (b) Formic acid (d) Ethanol (c) Acetic acid **28.** Which of the following microorganism is a true homolactic fermentor? (a) Propionobacterium (b) Staphylococcus (c) Streptococcus (d) Clostridium 29. Nitrogen fixation i.e. conversion of ammonia into nitrite and then to nitrate is carried out by (a) Nitrosomonas and Nitrobacter

4		\mathcal{E} t G - Olympiad Explorer
	(b) Nitrosomonas(d) Thiobacillus and Azotobact	(c) Nitrobacter er
30.		robic bacteria ? (b) <i>Nitrobacter</i> (d) All of the above.
31.	Common out breaks of food poi (a) Salmonella typhimurium (c) Clostridium nigrificans	
32.	fertility improvement? (a) Scytonema	organisms are recently used for soil (b) Anabaena (d) None of the above
33.		known to produce (b) Vitamin B (d) Vitamin K
34.		eed by (b) <i>Penicillium illudens</i> (d) <i>Ustilago illudens</i>
35.		nced by (b) P. glacum (d) P. griseofulvin
36.	species of Streptomyces? (a) S. rimosus, S. nodosus	antibiotics are produced by which two (b) S. fradiae, S. noursei (d) S. auerofaciens, S. erythraeus
37.		roduced by (b) <i>Bacillus subtilis</i> (d) <i>Fusarium orthoceras</i>
38.	(a) Lactobacillus	ommon souring and curding of milk is (b) Aspergillus (d) Clostridium
39.	and tea? (a) Penicillium glaucum	m is responsible for curing of tobacco (b) Micrococcus candisans (d) Mycoderma acti
40.	Retting of jute, flex and hemp fi one is responsible for retting?	bre is also enhanced by bacteria, which

(b) Penicillium notatum

(d) Micrococcus candisans

(a) Aspergillus nidulans

(c) Clostridium butyrium

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Class - 11
                                                                     5
41. Most of the eukaryotic organisms (except molds and yeasts) are
    (a) Gram neutral
                                   (b) Gram negative
    (c) Gram positive
                                   (d) None of the above
42. A toxoid is a
    (a) Toxin that loses its activity(b) Impotent toxin
    (c) Potent toxin
                                   (d) Toxin producing gland
43. Endotoxins are toxic chemicals secreted by pathogens inside the host
    cells, chemically they are
    (a) Lipids
                                   (b) Lipopolysaccharides
    (c) Proteins
                                   (d) Phospholipids
44. Lecithinase-causes lysis of RBC and other lipid containing tissues is
    produce by
    (a) Clostridium perfringenes
                                   (b) S. pvogenes
    (c) Streptococcus aureus
                                   (d) All of the above
45. The major difference between parasite and pathogen is that
    (a) One obtains his food only while other causes disease to the host
    (b) One causes disease and other fails to do so
    (c) Former is weaker where as latter is stronger in power
    (d) One lives inside while other lies outside the body
46. Virulence can be defined as
    (a) Ability of an organism to produce toxins
    (b) Toxicity caused by a parasite
    (c) Capacity of pathogen to cause disease
    (d) Ability of an organism to cause latent symptoms
47. Pseudomonas citri causes disease
                                   (b) Gall formation
    (a) Blight
    (c) Black spot
                                   (d) Citrus canker
48. 'White rust' disease of crucifer is caused by
    (a) Xanthomonas compestris (b) Ustilago spp
    (c) Albugo candida
                                   (d) All of the above
49. Sulfited spoilage of strawberry is caused by
    (a) Rhizopus stolonifer
                                   (b) R. sexualis
    (c) Both (a) and (b)
                                   (d) None of the above
50. Aspergillus candidus, A nidulus and A. niger are known to cause damage
    even to
    (a) Hardware of computer
                                   (b) Glassware
    (c) Optical instruments
                                   (d) All of these
51. Penicillin a potent antibiotic acts by destroying other microorganisms. It
```

(b) Cell wall formation

(d) RAN synthesis system

hinders

(a) Protein synthesis

(c) DNA formation

- **52.** Iodine is used as a sterilant, because it is a strong...... agent and destroys microorganisms
 - (a) Redox

(b) Reducing

(c) Oxidizing

- (d) All of these
- **53.** Which of the following growth curve is represented by bacterial cultures?
 - (a) J-shaped

(b) V-shaped

(c) S-shaped

(d) Linear shaped

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1. (d) 2. (b) 3. (a) 4. (a) 5. (a) 6. (a) 7. (a) 8. (b) 9. (c) 10. (d) 11. (c) 12. (d) 13. (b) 14. (c) 15. (b) 16. (d) 17. (a) 18. (c) 19. (b) 20. (a) 21. (b) 22. (c) 23. (d) 24. (b) 25. (c) 26. (c) 27. (a) 28. (c) 29. (a) 30.(e) 31. (a) 32. (a) 33. (b) 34. (a) 35. (c) 36. (b) 37. (d) 38. (a) 39. (b) 40. (c) 41. (b) 42. (a) 43. (b) 44. (a) 45. (a) 46. (c) 47. (d) 48. (c) 49. (a) 50. (a) 51. (b) 52. (c) 53. (a)

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EXERCISE-2

1.		tes can be compared to s (b) Mesosomes of prokaryotes s (d) Chromatophores of prokaryotes
2.	The most abundant element (a) Carbon(c) Hydrogen	found in living organisms is (b) Water (d) Nitrogen
3.	Which of the following gove (a) Nucleocytoplasmic ratio (c) Both (a) and (b)	
4.	The shape of cell may not b (a) Amoeba (c) Lactobacillus	e fixed (i.e. it changes) in (b) Chlamydomonas (d) Spirocheates
5.	Which of the following euka (a) Platelets (c) RBC	aryotic cell lacks nucleus (b) WBC (d) Nerve cell
6.	Which of the following representation (a) Qb (c) Bacteriophage	esents the smallest living organism ? (b) Amoeba (d) PPLO
7.	The major sterol found in et (a) Phytosterols (c) Ergosterol	ukaryotic microorganism cell membrane is (b) Cholesterol (d) Stigmasterol
8.	Cephalin is a (a) Phospholipid of bacteria (b) Phospholipid found in a (c) Phospholipid found in a (d) Glycolipid of bacterial a	ell membranes Il cell walls
9.	One of the following is a bit (a) Cartilage cells (b) Liver cells and cartilage (c) Liver cells (d) Parenchymatous cells	
10.	Which of the following org	ganelles are not present in prokaryotes? (b) DNA

(d) Endoplasmic reticulum

(b) Cell wall

(c) Ribosomes

(a) Nuclear membrane

11. Thin, elastic, selectively permeable membrane is

23. The term bioblast was used for mitochondria, by (a) Altman (b) Benda (c) A. Koliker (d) Porter 24. Mitochondria was first observed by (a) Altman [1892] (b) Benda [1898] (c) A. Koliker [1850] (d) Porter and Kallman 25. Smooth ER can be found in (a) Sarcoplasmic reticulum (b) Myeloid body (in retina of frog) (d) None of the above (c) Both of the above 26. Endoplasmic reticulum is absent in (a) Prokarvotes (b) Mature erythrocytes (c) Both of the above (d) None of these 27. Glycosylation of proteins occurs in (a) Endoplasmic reticulum (b) Mitochondria (c) Golgi body (d) All of the aese 28. Golgi body was first studied by Camello Golgi in (a) Cambial cells (b) Pus cells (c) Nerve cells of Brown owl (d) Muscle cells 29. Cardiopin is a membrane lipid, can be seen in (a) Bacteria (b) Chloroplast (c) Mitochondria (d) All of these **30.** Cell membrane is made up of (i) Phospholipids (ii) Proteins (iii) Glycolipids (iv) Sterols (v) Cellulose Choose the correct answer. (a) i, ii, iii, iv (b) i, iii, v (c) i, iii, iv, v (d) ii, iii, v 31. Photophosphorylation occurs in (a) Golgi body (b) Chloroplast (d) Cell membrane (c) Mitochondria **32.** ATP synthetase, PS-I, PS-II and glutamase dehydrogenase are characteristic of (a) Matrix (b) Thylakoid (c) Outer membrane (d) Inner membrane 33. Photo systems (PS-I and PS-II) are found in (a) Stroma (b) Thylakoid membrane (c) Outer membrane (d) Inner membrane **34.** Synthesis of starch, sugars and fatty acid occurs in

10

EtG Olympiad Explorer

(a) Granum

(b) Inner chloroplast membrane

(c) Thylakoid membrane

(d) Stroma

35. The term 'plastid' was first used by

(a) Schimper

(b) Hoffman

(c) Menke

(d) Altman

- **36.** Mitochondria is known as 'power house of cell', because
 - (a) It liberates heat

(b) It stores energy

(c) It is hot

(d) It produces ATPs

- 37. Cytochrome oxidase and cytochrome deficiency in mitochondria causes
 - (a) Kearns-Says syndrome
 - (b) Kearns-Says syndrome and Menke's disease respectively.
 - (c) Menke's disease
 - (d) Leber's optic neuropathy
- **38.** Citrate synthetase, succinate dehydrogenase, fatty acyl coA synthetase are common enzymes of
 - (a) Matrix
 - (b) Inner mitochondrial membrane
 - (c) Outer mitochondrial membrane
 - (d) Perimitochondrial space
- 39. Inner mitochondrial membrane is rich in
 - (a) Choline phosphotransferase
 - (b) Sulphite oxidase
 - (c) Cation translocases/fatty acyl transferases
 - (d) Citrate synthetase

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1. (b) 2. (a) 3. (a) 4. (a) 5. (c) 6. (d) 7. (c) 8. (a) 9. (b) 10. (d) 11. (c) 12. (c) 13. (b) 14. (c) 15. (d) 16. (c) 17. (b) 18. (a) 19. (c) 20. (c) 21. (d) 22. (a) 23. (a) 24. (c) 25. (a) 26. (a) 27. (c) 28. (c) 29. (d) 30. (a) 31. (b) 32. (b) 33. (b) 34. (d) 35. (a) 36. (d) 37. (b) 38. (a) 39. (c)

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EXERCISE-3

1.	(a) 37°C (c) 25°C	at which assays are performed are (b) 30°C (d) All of these
2.	A heterolytic fission of enzyme (a) Carbanion and Carbonium (b) Carbonium ion (c) Carbanion	e bonds results in the formation by ion (d) Free radical
3.	A homolytic fission of enzyme (a) Carbonium ion(c) Carbanion	bonds results in the formation of (b) Free radical (d) Cations
4.	The inhibitor and the enzyme (a) Covalent bond (c) Electrostatic	usually bind by (b) Tandons force (d) Vander waal force
5.	The region which contains the as	e binding and catalytic sites is termed
	(a) Active region(c) Active site	(b) Active centre(d) All of these
6.	Enzymes are (a) Globular proteins (c) Fibrous proteins	(b) Glycoproteins(d) Lipoproteins
7.	Enzymes lower the activation of (a) Binding energy (c) ATP	energy of reaction by using (b) GTP (d) None
8.	Enzymes affect (a) Reaction rates & reaction (b) Reaction rates and equilibrate (c) Reaction rates (d) Substrate and product conditions	ria
9.	Lyases catalyzes reaction around (a) Co-ordinate bond	d (b) Double bond
	(c) Single bond	(d) All of these
10.	If a complex organic or metallo it is called as (a) Apoprotein	organic group is attached to an enzyme (b) Coenzyme
	(c) Cofactor	(d) Prosthetic group

11. Which of the following statements is wrong?

		COC Clympiaa Explorer
	(a) Some DNA molecules have(b) Few RNA molecules behave(c) Some RNA molecules may(d) Enzymes are the largest a molecules	ve like enzyme
12.	•	d and free enzymes are the same enzyme changes when is in free solution
13.	Antibody that behaves as enzyr (a) Lipozymes (c) Isozymes	nes in catalyzing reactions are called (b) Holozymes (d) Abzymes
14.	Abzymes are (a) Antibody (c) Hydrolases	(b) Antigen(d) Proteases
15.	Enzymes obtained commercially (a) Intracellular(c) Extracellular	from microbial fermentation is usually (b) Intercellular (d) Acellular
16.	Extracellular enzymes are those (a) Outside the cell(c) In between two cells	that are present (b) Periplasmic space (d) Inside the cell
17.	Coenzymes are preferably store (a) In sodium sulphate(c) As solid	d a (b) As liquid (d) In liquid nitrogen
18.	In general enzymes are stored a (a) 20°C (c) 60°C	at (b) 120°C (d) 40°C
19.	Enzyme-catalyzed reactions are (a) They proceed under relativ (b) They are easily available (c) They are resistant to high a (d) They proceed with a great	and low temperature
20.	Enzyme activity in terms of tot (a) Molar activity (c) Molar catalytic activity	(b) Specific activity
21.	Out of the total enzymes present (a) 4 % (c) 95 %	nt in the cell, mitochondria alone has (b) 70 % (d) No enzyme

22.	(a) Lipase	est the food reserve in castor seeds (b) Diastase (d) Proteases
23.	Enzymes differ from inorganic c (a) Not being used up in reacti (b) Working at high temperature (c) Having high diffusion rate	on e
24.	` / 1	s is active at acidic pH ? (b) Amylase (d) Chymotrypsin
25.		e systems are associated with (b) Biological membrane (d) None of these
26.		(b) Intracellular space (d) Nucleolus.
27.	•	of enzymes. (b) Isoenzymes (d) Neoenzymes
28.	•	y cells are (b) Stored (d) Neoenzymes
29.	(a) Michaelis and Menten	at the active site was postulated by (b) Sutton (d) Franklin
30.		an be (b) Double (d) Half of the total active site
31.	() 88	n (b) Human vacuole (d) Lysosome
32.	* *	(b) Gibbs free energy (d) Order of reaction
33.	` / •	s are not monomeric enzymes ? (b) Chymotrypsin (d) Lactose synthetase
34.	Monomeric enzymes are those tl	hat are made up of

- (a) Straight chain polypeptides (b) Single polypeptide chain
- (c) Single type of amino acid (d) All of these
- 35. Peptide linkage occurs between
 - (a) Sucrose

- (b) Glucose
- (c) Amino acids
- (d) Glycogen
- **36.** Which is not an attribute of enzyme
 - (a) They are specific in nature
 - (b) They speed up rate of biochemical reaction
 - (c) They are used up in reaction
 - (d) All of the above
- **37.** NADP is
 - (a) A coenzyme
- (b) A part of s-RNA
- (c) An enzyme
- (d) A part of t-RNA
- **38.** Dry seeds can endure higher temperature than soaked germinating seeds because
 - (a) Dry seeds have more reserved food
 - (b) Seedlings are tender
- (c) Dry seeds are hard
- (d) Hydration makes the enzyme sensitive to temperature
- **39.** Diastase enzyme digests
 - (a) Fat

(b) Proteins

(c) Starch

- (d) Cellulose
- 40. Esterase belongs to
 - (a) Hydrolytic enzyme
- (b) Carboxylating enzyme
- (c) Oxidation reduction enzyme
- (d) Transferases

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1.	(a)	2. (a)	3. (b)	4. (a)	5. (a)	6. (a)	7. (a)	8. (a)
9.	(b)	10. (b)	11. (a)	12. (b)	13. (d)	14. (a)	15. (c)	16. (a)
17.	(c)	18. (d)	19. (a)	20. (b)	21. (b)	22. (a)	23. (d)	24. (a)
25.	(a)	26. (b)	27. (a)	28. (d)	29. (a)	30. (a)	31. (d)	32. (a)
33.	(a)	34. (b)	35. (c)	36. (c)	37. (a)	38. (d)	39. (c)	40. (a)

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EXERCISE-4

1.	(a)	mbrane proteins are Aligned diagonally Symmetrically placed		Asymmetrically placed Arranged in a zig zag manner
2.	(a)	ich of the following protein tr Glycophorin Spectrin	(b)]	ses the bilayer as β helix ? Porin None of these
3.	(a)	ctrin is attached to the member Hydrogen bond Phosphodiester bond	(b) .	through Ankyrin Covalent bond
4.	(a)	cophorin extends the Red blo Single α helix Single β helix	(b) 1	ll lipid bilayer as Double β helix Triple α helix
5.	(a)	cophorin is a Cytosolic protein Transmembrane protein		Peripheral protein Triple α - helix
6.	(a)	bohydrate chain is attached to 16 residues Varies between 16-17	(b)	cophorin at 17 residues 26 residues
7.		proteins in the membrane ex Rotational movement Transverse movement	(b)]	Lateral movement No movement
8.	(a)(b)(c)	Hydrogen bonds between pe helix	ptide	ation is always an-helical because bonds are maximum in case of an any other conformation of prote
9.	(a)	kbone of majority of phospho Alcohol Glycerol	(b)]	s are made up of : Fatty acid Sphingosine
10.	(a)	lipid molecules and proteins Non covalent bond Hydrogen bond	(b) .	embranes are held together by Any covalent bond Electrostatic force
11.	An	ATPase pump that does not ta	ke pl	lace in eukaryotes is

(b) Ca²⁺ pump

(d) None

(a) K⁺ pump

(c) Na⁺-K⁺ pump

_		COC Olympiaa Explorer
12.	In case of bacteria the carrier pro (a) Both (c) H ⁺ gradient only	teins pump variety of nutrients by using (b) ATPase only (d) None
13.	Ion channels always remain (a) Fluctuate between open and (b) Open (d) Some are always closed and	(c) Closed
14.	Channel proteins that allow ion communicating cells are (a) Gap junctions (c) Na ⁺ - K ⁺ pump	ns and small molecules to flow between (b) Ca ²⁺ pump (d) All of these
15.	Which of the following ions bind (a) Al (c) K ⁺	l strongly to valinomycin ? (b) Na ⁺ (d) Ca
16.	Ionophores are (a) Neutral molecules (c) Hydrophillic molecules	(b) Hydrophobic molecules(d) Positively charged
17.	The transport by channel protein (a) Active (c) Both	s is (b) Passive (d) None
18.	The transport by carrier proteins (a) Active(c) Both	is (b) Passive (d) None
19.	Active transport is mediated by (a) Both depending on the conc (b) Carrier protein (c) Channel protein (d) Both depending on the size 	
20.	The ions traverse through the me (a) G-protein(c) Channel protein	embrane faster through (b) Carrier protein (d) Free lipid bilayer
21.	Ras proteins contain subfamilies(a) Rab proteins(c) Rho proteins	The one which is not a part of it are(b) Rac proteins(d) SH₂ proteins
22.	Which of the following receptor (a) G-protein(c) Both	used CGMP as intracellular mediator? (b) Guanyly cyclases (d) None
23.	Which of the following is not a (a) Receptor tyrosine phosphata	

(b) Receptor tyrosine kinases

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(c) Receptor guanyl cycases (d) None of the above
24. In case of vertebrate vision the cyclic nucleotide involved is
    (a) cCMP
                                    (b) cGMP
    (c) cAMP
                                    (d) cTMP
25. In a cell of vertebrate vision the cyclic nucleotide involved is
    (a) 24 connexin
                                    (b) 12 connexin
                                    (d) 14 connexin
    (c) 10 connexin
26. Which of the following is not an energy source in active transport?
    (a) Sodium motive force
                                    (b) Proton motive force
    (c) ATP & Phosphoenol phruvate
    (d) All of the above
27. Vitamin D is derived from
    (a) Mevalonate
                                    (b) Light
    (c) Cholesterol
                                    (d) HMG CoA
28. The basis structural unit of sphingolipids is
    (a) Palmitoyl CoA
                                    (b) Sphingosine
    (c) Ceramide
                                   (d) Dehydrosphinganine
29. A phospholipid which is formed from dihydroxy acetone phosphate (DAP)
    is
                                    (b) Ether phospholipids
    (a) Choline phospholipid
    (c) Glycolipids
                                    (d) Phospholipid
30. G-Protein is
                                    (b) Bimeric
    (a) Trimeric
    (c) Unimeric
                                    (d) Tetrameric
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Class - 11

1. (c) 2. (a)

9. (c) 10. (a) 11. (d)

17. (b) 18. (b) 19. (a)

3. (b)

25. (b) 26. (d) 27. (c) 28. (c) 29. (c) 30. (a)

4. (b)

5. (a) 6. (b)

12. (a) 13. (a) 14. (a)

20. (b) 21. (d) 22. (b)

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7. (c)

15. (c) 16. (b)

23. (a) 24. (b)

8. (a)

NATIONWIDE BIOTECHNOLOGY OLYMPIAD (NBTO) SAMPLE PAPER

Total Marks: 50 **Total duration :** 60 Minutes

	SECTION A: GENERAL KNOWLEDGE
1.	How many years does it take glass to decompose? (a) 100 years or less (b) 100 to 500 years (c) 500 to 1000 years (d) Over 1 million years
2.	Select the appropriate SEQUENCE. (a) Global warming → melting snow → rising sea level floods (b) Volcanic dust → traps carbon dioxide → global warming (c) Greenhouse effect → carbon dioxide traps heat → melting snow (d) All of the above
3.	What human activities cause ozone depletion? (a) Industrial halocarbons (b) Smog from vehicles (c) Cigarettes (d) Cooking
4.	"Close the loop on recycling" is becoming a popular catchphrase in the world of waste management. What does it refer to? (a) Buying recycled products (b) Dropping off recycled materials at a certified recycling center (c) Living an environmental lifestyle besides just recycling (d) Tying up your recyclables securely
5.	What recyclable commodity is most valuable? (a) Paper (b) Plastic (c) Glass (d) Aluminum
6.	 The major threats to biodiversity worldwide can be categorized using the HIPPO dilemma. HIPPO stands for: (a) Hippopotamuses (b) Habitat Loss, Introduced Species, Pollution, Population, Overconsumption (c) Hunting, Isolation, People, Propaganda, Opulence (d) Harmful ultraviolet rays, International trade, Politics, Power production, Oxygen deprivation
7.	Speaking of the ozone layer, what harmful product does this part of the atmosphere filter? (a) Greenhouse Effect (b) Acid Rain (c) UV light (d) Sewage
8.	What gas has been continuously highlighted as being the main contributor to global warming? (a) Carbon monoxide (b) Sulphur dioxide (c) Methane gas (d) Carbon dioxide
9.	What do scientists believe global warming is caused from? (a) The earth getting older (b) Human activity (c) Changes in animals (d) The rainforest

Class - 11

- 10. Scientists are worried that if the Greenland ice cap melts rapidly, it will deliver a surge of freshwater into the North Atlantic Ocean. This will cause what to happen?
 - (a) Icy climates will occur in that region.
 - This freshwater will shutdown the ocean conveyor in this region.
 - The warmth that is delivered to Europe via the Gulf Stream leg of the conveyor will cease in activity.
 - (d) All of the above.

SECTION B: PHYSICS & CHEMISTRY

- 11. Two masses $m_1 = 5$ kg and $m_2 = 4.8$ kg tied to a string are hanging over a light frictionless pulley. What is the acceleration of the masses when lift is free to move? $(g = 9.8 \text{ m/s}^2)$
 - (a) 0.2 m/s^2
 - (b) 9.8 m/s^2
 - (c) 5 m/s^2
 - (d) 4.8 m/s^2



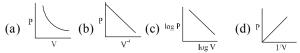
- 12. If g is the acceleration due to gravity on the earth's surface, the gain in the potential energy of an object of mass m raised from the surface of the earth to a height equal to the radius R of the earth, is
- (a) 2mgR (b) $\frac{1}{2}mgR$ (c) $\frac{1}{4}mgR$
- 13. A wire fixed at the upper end stretches by length l by applying a force F. The work done in stretching is
- (b) F l (c) 2F l
- 14. In an LCR series ac circuit, the voltage across each of the components. L, C and R is 50 V. The voltage across the LC combination will be
 - (a) 50 V
- (b) $50\sqrt{2}V$
- (c) 100 V
- (d) 0 (zero)
- 15. An α -particle of energy 5 MeV is scattered through 180° by a fixed uranium nucleus. The distance of the closest approach is of the order of
 - (a) 1Å
- (b) 10^{-10} cm (c) 10^{-12} cm
- (d) 10^{-15} cm
- 16. When the velocity of electron increases, its specific charge
 - (a) decreases
- (b) increases
- (c) remain unaffected (d) none
- **17.** In the reaction

$$KMnO_4 + H_2SO_4 + H_2C_2O_4 \longrightarrow product$$

Mn²⁺ ions act as

- (a) positive catalyst (b) negative catalyst
- (c) autocatalyst
- (d) enzyme catalyst.

- 18. Which of the following compounds on boiling with alkaline KMnO₄ and subsequent acidification will not give benzoic acid?
 - (a) anisole
- (b) benzyl alcohol
- (c) toluene
- (d) acetophenone
- 19. The benzyl phenyl ether on reaction with HI gives
 - (a) benzyl iodide
- (b) benzyl alcohol
- (c) benzyl iodide & phenol
- (d) benzyl alcohol & iodobenzene.
- **20.** Which one is not a graphical representation of Boyle's law?



SECTION C: MATHEMATICS

- 21. $\lim_{x\to\infty} \frac{(x+1)^{10} + (x+2)^{10} + \dots + (x+100)^{10}}{x^{10} + 10^{10}}$ is equal to
 - (a) 0
- (b) 1
- (c) 100
- 22. $\lim_{n\to\infty} \left[\frac{n}{1+n^2} + \frac{n}{4+n^2} + \frac{n}{9+n^2} + \dots + \frac{1}{2n} \right]$ is equal to

- (a) $\frac{\pi}{2}$ (b) $\frac{\pi}{4}$ (c) 1 (d) none of these
- 23. If $\vec{A} = i + 2j + 3k$, $\vec{B} = -i + 2j + k$ and $\vec{C} = 3i + j$, then the value of t such that $\vec{A} + t \vec{B}$ is at right angle to vector \vec{C} , is (a) 2 (b) 4 (c) 6 (d) 5
- 24. The principal value of $\sin^{-1}\left(\sin\frac{5\pi}{3}\right)$ is
 - (a) $\frac{5\pi}{3}$ (b) $-\frac{\pi}{3}$ (c) $-\frac{5\pi}{3}$ (d) $\frac{4\pi}{3}$

- 25. If the lines represented by the equation $2x^2 3xy + y^2 = 0$ make angles α and β with x-axis, then $\cot^2 \alpha + \cot^2 \beta =$
- (a) 0 (b) $\frac{3}{2}$ (c) $\frac{5}{4}$ (d) $\frac{7}{4}$
- **26.** Range of the function $\frac{2}{2-\sin 3x}$ is
- (a) [1,3] (b) $\left(\frac{1}{3},1\right)$ (c) (1,3) (d) $\left[\frac{1}{3},1\right]$

- 27. If a, b, c, d are positive real numbers, then $\lim_{n\to\infty} \left(1 + \frac{1}{a+bn}\right)^{c+dn}$ is equal to
- (a) $e^{\frac{d}{b}}$ (b) $e^{\frac{c}{a}}$ (c) $e^{\frac{(c+d)}{a(a+b)}}$ (d) $e^{\frac{c}{a}}$
- **28.** If $\int_{0}^{1} t^2 f(t) dt = (1 \sin x)$, then $f\left(\frac{1}{\sqrt{3}}\right)$ is equal to
- (a) $\frac{1}{3}$ (b) 3 (c) $\frac{1}{\sqrt{3}}$ (d) $\sqrt{3}$
- **29.** $\tan^{-1}x + \tan^{-1}y + \tan^{-1}z = \pi$, then $\frac{1}{xy} + \frac{1}{yz} + \frac{1}{zx}$ is equal to

 - (a) xyz (b) $\frac{1}{xyz}$ (c) 1 (d) 0
- **30.** If for any complex number z, |z-4| < |z-2|, then
 - (a) Re(z) > 0 (b) Re(z) < 0
- - (c) Re(z) > 3
- (d) Re(z) > 2
- **31.** If $(1 + \omega)^7 = A + B\omega$, then A, B are

 - (a) 0, 1 (b) 1, 1 (c) 1, 0
- (d) -1, 1
- **32.** The roots of $|x-2|^2 + |x-2| 6 = 0$ are

 - (a) 0.4 (b) -1.3 (c) 4.2
- (d) 5, 1
- **33.** If two sets A and B are having 99 elements in common, then the number of elements common to each of the sets $A \times B$ and $B \times A$ are (c) 100 (d) 18
 - (a) 99^2
- (b) 2^{99}
- 34. The order and degree of the differential equation representing family of curves $y^2 = 2k \left(x + \sqrt{k}\right)$ (where k is a positive parameter) are
- (a) 1,2 (b) 1,3 (c) 1,4
- (d) 2, 4
- 35. The integrating factor of the differential equation
 - $\frac{dy}{dx}(x \log x) + y = 2 \log x$ is
- (b) e^x (c) $\log(\log x)$ (d) $\log x$
- **36.** The system of equations x + y + z = 2, 2x + y z = 3, 3x + 2y + kz = 4 has unique solution if
 - (a) k = 0 (b) -2 < k < 2 (c) -1 < k < 1 (d) $k \ne 0$
- 37. If $A = \begin{bmatrix} 2 & -3 \\ 1 & -1 \end{bmatrix}$, then $|A^{1003} 5A^{1002}|$ is equal to
 (a) 1 (b) 18 (c) 21 (d) -5
- **38.** If $(1+x)^n = C_0 + C_1 x + C_2 x^2 + \dots + C_n x^n$, then $C_0 C_r + C_1 C_{r+1} + C_2 C_{r+2} + \dots + C_{n-r} C_n$, is equal to

- (a) $\frac{(2n)!}{(n-r)!(n+r)!}$ (b) $\frac{(2n)!}{n!(n+r)!}$

- (c) $\frac{(2n)!}{n!(n-r)!}$ (d) $\frac{(2n)!}{(n-1)!(n+1)!}$
- 39. tan1° tan2° tan3°.....tan89° is equal to

Class - 11

- (a) 0 (b) 1 (c) ∞
- (d) none of these
- **40.** If $x^2 + \alpha y^2 + 2\beta y = a^2$ represents a pair of perpendicular stright lines,

 - (a) $\alpha = -1, \beta = +a$ (b) $\alpha = 1, \beta = a$
 - (c) $\alpha = -1, \beta = 0$ (d) None of these

SECTION C: BIOTECHNOLOGY

- 21. Diagrams, tables, and graphs are used by scientists mainly to
 - (a) design a research plan for an experiment
 - (b) test a hypothesis
 - (c) organize data
 - (d) predict the independent variable
- 22. What happens to certain nutrient molecules after they pass into muscle cells?
 - (a) They are replicated in the nucleus.
 - (b) They are acted on by enzymes and release the energy they contain.
 - (c) They are changed into tissues and organs in the cytoplasm.
 - (d) None of these
- 23. A medical test indicates that a patient has a defective protein. This condition is most likely due to a change in the directions coded in
 - (a) number of hydrogen atoms in starch molecules
 - (b) sequence of inorganic molecules
 - (c) number of carbon atoms in sugar molecules
 - (d) sequence of subunits in DNA
- 24. If a human system fails to function properly, what is the most likely result?
 - (a) a stable rate of metabolism
 - (b) a disturbance in homeostasis
 - (c) a change in the method of cellular respiration
 - (d) a change in the function of DNA
- 25. Which statement regarding the functioning of the cell membrane of all organisms is *not* correct?

39

- (a) The cell membrane forms a boundary that separates the cellular contents from the outside environment.
- (b) The cell membrane is capable of receiving and recognizing chemical signals.
- (c) The cell membrane forms a barrier that keeps all substances that might harm the cell from entering the cell.
- (d) The cell membrane controls the movement of molecules into and out of the cell.
- **26.** In multicellular organisms, cells must be able to communicate with each other. Structures that enable most cells to communicate with each other are known as
 - (a) pathogenic agents
- (b) chloroplasts
- (c) antibiotics
- (d) receptor molecules
- 27. Every single-celled organism is able to survive because it carries out
 - (a) metabolic activities (b) autotrophic nutrition
 - (c) heterotrophic nutrition(d) sexual reproduction
- 28. The shape of a protein molecule is influenced by
 - (a) whether it is organic or inorganic
 - (b) the sequence of amino acids in it
 - (c) the number of genes found in the nucleus
 - (d) the number of chromosomes in the cell
- 29. A small amount of DNA was taken from a fossil of a mammoth found frozen in glacial ice. Genetic technology can be used to produce a large quantity of identical DNA from this mammoth's DNA. In this technology, the original DNA sample is used to
 - (a) stimulate differentiation in other mammoth cells
 - (b) provide fragments to replace certain human body chemicals
 - (c) act as a template for repeated replication
 - (d) trigger mitosis to obtain new base sequences
- 30. Many diabetics are now using insulin that was made by certain bacteria. The ability of these bacteria to produce insulin was most likely the result of
 - (a) deleting many DNA segments from bacterial DNA
 - (b) genetic mapping of bacterial DNA to activate the gene for insulin production
 - (c) inserting a portion of human DNA into the ring-shaped DNA of bacteria
 - (d) using radiation to trigger mutations
- 31. Which situation would most directly affect future generations naturally produced by a mango tree?
 - (a) Ultraviolet radiation changes the DNA sequence within some leaves of the tree.

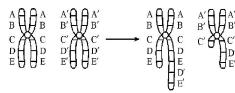
- (b) Ultraviolet radiation changes the DNA sequence within the gametes of some flowers of the tree.
- (c) An increase in temperature reduces the number of cell divisions in the roots.
- (d) Rapidly growing cells just under the bark are exposed to radiation, causing changes in genetic material.
- **32.** Which statement is best supported by fossil records?
 - (a) Many organisms that lived in the past are now extinct.
 - (b) Species occupying the same habitat have identical environmental needs.
 - (c) The struggle for existence between organisms results in changes in populations.
 - (d) Structures such as leg bones and wing bones can originate from the same type of tissue found in embryos.
- 33. One explanation for the variety of organisms present on Earth today is that over time
 - (a) new species have adapted to fill available niches in the environment
 - (b) evolution has caused the appearance of organisms that are similar to each other
 - (c) each niche has changed to support a certain variety of organism
 - (d) the environment has remained unchanged, causing rapid evolution
- 34. Within which structure in the human body does specialization of parts of the developing baby take place?
 - (a) ovary

- (b) uterus (c) testis (d) pancreas
- 35. Which statement best explains the significance of meiosis in the process of evolution within a species?
 - (a) The gametes produced by meiosis ensure the continuation of any particular species by asexual reproduction.
 - (b) Equal numbers of eggs and sperm are produced by meiosis.
 - (c) Meiosis produces eggs and sperm that are alike.
 - (d) Meiosis provides for variation in the gametes produced by an organism.
- **36.** Which phrase best describes cellular respiration, a process that occurs continuously in the cells of organisms?
 - (a) removal of oxygen from the cells of an organism
 - (b) conversion of light energy into the chemical bond energy of organic molecules
 - (c) transport of materials within cells and throughout the bodies of multicellular organisms
 - (d) changing of stored chemical energy in food molecules to a form usable by organisms

- (a) They are produced by the body in response to the presence of foreign substances.
- (b) They may be produced in response to an antigen.
- (c) They are nonspecific, acting against any foreign substance in the body.
- (d) They may be produced by white blood cells.
- **38.** An increase in the level of insulin in the blood would most directly result in
 - (a) a decrease in the amount of glucose in the blood
 - (b) a decrease in the amount of protein in the blood
 - (c) an increase in the amount of fat in cells
 - (d) an increase in the amount of carbon dioxide in cells
- **39.** Compared to a natural forest, the wheat field of a farmer *lacks*
 - (a) heterotrophs
- (b) significant biodiversity
- (c) autotrophs
- (d) stored energy
- **40.** A new type of fuel gives off excessive amounts of smoke. Before this type of fuel is widely used, an ecologist would most likely want to know
 - (a) what effect the smoke will have on the environment
 - (b) how much it will cost to produce the fuel
 - (c) how long it will take to produce the fuel
 - (d) if the fuel will be widely accepted by consumers

SECTION D: INTERACTIVE QUESTIONS

41. The diagram below represents a change that occurred in a pair of chromosomes during the formation of an egg cell. The letters represent genes on the pair of chromosomes.



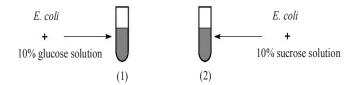
The alteration that occurred will most likely

- (a) be passed on to every cell that develops from the egg cell
- (b) change the chromosome number of the body cells that develop from the egg cell
- (c) convert sex cells into body cells
- (d) trigger the production of pathogens
- **42.** The diagram below represents chromosomes in a zygote.



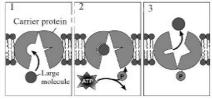
Which diagrams best illustrate the daughter cells that result from normal mitotic cell division of this zygote?

- **43.** The diagram below shows two setups that were used to study bacterial growth. Each setup initially contained an equal number of the bacterium *E. coli* in different carbohydrate solutions. After one hour, a 1-milliliter sample was drawn from each tube and analyzed. The number of bacteria found in the sample from test tube 1 was higher than the number in test tube 2.



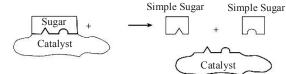
Which conclusion regarding this investigation is *not* valid?

- (a) All bacteria grow best in a solution of glucose.
- (b) *E. coli* grows better in a 10% solution of glucose than in a 10% solution of sucrose.
- (c) The type of sugar solution will make a difference in the rate of growth of *E. coli*.
- (d) The rate of growth of *E. coli* depends on the type of carbohydrate present.
- **44.** The diagram below represents movement of a large molecule across a membrane.



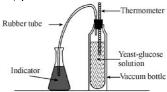
Which process is best represented in this diagram?

- (a) active transport
- (b) diffusion
- (c) protein building
- (d) gene manipulation
- **45.** The diagram below illustrates a biochemical process that occurs in organisms.



The substance labeled "catalyst" is also known as

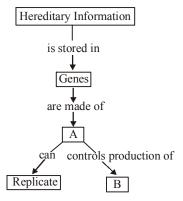
- (a) a hormone
- (b) an antibody
- (c) an enzyme
- (d) an inorganic compound
- **46.** A student placed a solution of glucose and yeast in a vacuum bottle and sealed it with a two-hole stopper as shown in the diagram below. The temperature of the yeast glucose solution increased gradually with time, and the colour of the indicator was observed and recorded throughout a 2-day period.



The purpose of the investigation was most likely to

- (a) study the relationship between temperature and pressure
- (b) demonstrate the release of energy by a chemical process
- (c) show that proteins are produced by yeast
- (d) study autotrophic nutrition in yeast

Base your answers to questions 47 and 48 on the diagram below, which provides information related to heredity.



- 47. The type of molecule in box A serves as a template. This means
 - (a) It serve as complete genome
 - (b) It serve as a pattern
 - (c) It serve as sugar molecule
 - (d) None of these
- **48.** Which molecules are represented by box *B*?
 - (a) bases

- (b) proteins
- (c) amino acids
- (d) simple sugars

Base your answers to questions 49 and 50 on the information below.

A student completed a series of experiments and found that a protein digesting enzyme (intestinal protease) functions best when the pH is 8.0 and the temperature is 37°C. During an experiment, the student used some of the procedures listed below.

Procedures

- (A) Adding more protease
- (B) Adding more protein
- (C) Decreasing the pH to 6.0
- (D) Increasing the temperature to 45°C
- (E) Decreasing the amount of light
- **49.** Which procedure would have the *least* effect on the rate of protein digestion?
 - (a) A
- (b) *E*
- (c) C
- (d) D
- **50.** Which two procedures would most likely cause a *decrease* in the rate of protein digestion?
 - (a) *A* and *D*
- (1
- (b) B and C (c) C and D
- (d) A and E

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◎ END OF THE EXAM ◎

		ANSWER	S	
1. (d)	2. (a)	3. (a)	4. (a)	5. (d)
6. (b)	7. (c)	8. (d)	9. (b)	10. (d)
11. (a)	12. (b)	13. (d)	14. (d)	15. (c)
16. (a)	17. (c)	18. (a)	19. (c)	20. (b)
21. (c)	22. (b)	23. (d)	24. (b)	25. (c)
26. (d)	27. (a)	28. (b)	29. (c)	30. (c)
31. (b)	32. (a)	33. (a)	34. (d)	35. (d)
36. (d)	37. (c)	38. (a)	39. (b)	40. (a)
41. (a)	42. (b)	43. (a)	44. (a)	45. (c)
46. (b)	47. (b)	48. (b)	49. (b)	50. (c)

